

Post-Lab Questions

1. Compare the results for your $[\text{Cu}^{2+}]$ unknown from the Daniell cell and the concentration cell:
 - (a) What is the percent difference?

 - (b) Which do you think is more accurate? Why?
2. If the temperature in the lab *increased* as you progressed through your dilutions for the Daniell cell investigation, how would this affect the slope and intercept of your E_{cell} vs. $\ln(1/[\text{Cu}^{2+}])$ plot?
3. Calculate the theoretical potential of a Daniell cell prepared with solutions of $[\text{Cu}^{2+}] = 0.50 \text{ M}$ and $[\text{Zn}^{2+}] = 0.05 \text{ M}$.
4. Instead of using Cu^{2+} and a Cu(s) electrode in the cathode half-cell, you use a solution of Sn^{2+} and a Sn(s) electrode. How would this affect the slope and intercept of your plot?

Lab 10: Electrochemical Cells and the Nernst Equation

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