

Name: KEY Date: \_\_\_\_\_

**Balancing Equations**

(2)

Balance the following chemical equations.

- $\underline{\quad}$  CH<sub>4</sub> +  $\underline{2}$  O<sub>2</sub> →  $\underline{\quad}$  CO<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>O
- $\underline{\quad}$  Na<sup>+</sup> +  $\underline{\quad}$  Cl<sup>-</sup> →  $\underline{\quad}$  NaCl
- $\underline{4}$  Al +  $\underline{3}$  O<sub>2</sub> →  $\underline{2}$  Al<sub>2</sub>O<sub>3</sub>
- $\underline{\quad}$  N<sub>2</sub><sup>+</sup> +  $\underline{3}$  H<sub>2</sub> →  $\underline{2}$  NH<sub>3</sub>
- $\underline{8}$  CO(g) +  $\underline{17}$  H<sub>2</sub>(g) →  $\underline{\quad}$  C<sub>8</sub>H<sub>18</sub>(l) +  $\underline{8}$  H<sub>2</sub>O
- $\underline{\quad}$  Fe<sub>2</sub>O<sub>3</sub>(s) +  $\underline{3}$  CO(g) →  $\underline{2}$  Fe(l) +  $\underline{3}$  CO<sub>2</sub>(g)
- $\underline{2}$  H<sub>2</sub>SO<sub>4</sub> +  $\underline{\quad}$  Pb(OH)<sub>4</sub> →  $\underline{\quad}$  Pb(SO<sub>4</sub>)<sub>2</sub> +  $\underline{4}$  H<sub>2</sub>O
- $\underline{2}$  Al +  $\underline{6}$  HCl →  $\underline{2}$  AlCl<sub>3</sub> +  $\underline{3}$  H<sub>2</sub>
- $\underline{\quad}$  Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>SO<sub>4</sub> →  $\underline{2}$  CaSO<sub>4</sub> +  $\underline{\quad}$  Ca(H<sub>2</sub>PO<sub>4</sub>)<sub>2</sub>
- $\underline{\quad}$  H<sub>3</sub>PO<sub>4</sub> +  $\underline{5}$  HCl →  $\underline{\quad}$  PCl<sub>5</sub> +  $\underline{4}$  H<sub>2</sub>O

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